

Periodic Table of the Elements

1
IA

1 1.01
H 1+
hydrogen
2
IIA

Average relative masses are 2001 values, rounded to two decimal places. All average masses are to be treated as measured quantities, and subject to significant figures rules. Do not round them further when performing calculations.

2 1.0
Li 1+
lithium
3 1.5
Be 2+
beryllium

11 22.99
Na 1+
0.9
sodium
12 24.31
Mg 2+
1.2
magnesium

19 39.10
K 1+
0.8
potassium
20 40.08
Ca 2+
1.0
calcium

37 85.47
Rb 1+
0.8
rubidium
38 87.62
Sr 2+
1.0
strontium

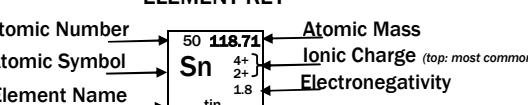
55 132.91
Cs 1+
0.7
cesium
56 137.32
Ba 2+
0.9
barium

87 [223]
Fr 1+
0.7
francium
88 [226]
Ra 2+
0.9
radium

3 IIIB
4 IVB
5 VB
6 VIB

7 VIIIB
8
9
10
11 IB
12 IIB

ELEMENT KEY



Elements 43, 61, and 84 through 118 are unstable and radioactive. The most stable isotope is listed inside parentheses for radioactive elements.

13 IIIA
14 IVA
15 VA
16 VIA
17 VIIA

5 10.81
B 3+
2.0
boron
6 12.01
C 4+
2.5
carbon
7 14.01
N 3-
3.0
nitrogen
8 16.00
O 2-
3.5
oxygen
9 19.00
F 1-
4.0
fluorine

10 20.18
Ne
helium
13 26.98
Al 3+
1.5
aluminum
14 28.09
Si 4+
1.8
silicon
15 30.97
P 3-
2.1
phosphorus
16 32.07
S 2-
2.5
sulfur
17 35.45
Cl 1-
3.0
chlorine
18 39.95
Ar
argon

31 69.72
Ga 4+
1.8
germanium
32 72.61
Ge 2-
2.4
selenium
33 74.92
As 3-
2.0
arsenic
34 78.96
Se 2-
2.8
bromine
35 79.90
Br 1-
3.0
krypton

36 83.80
Kr
3.0
36 127.60
Te 2-
2.1
tellurium
37 126.90
I 1-
2.5
iodine
38 131.29
Xe 2.6
xenon

50 118.71
Sn 3+
2.0
tin
51 121.76
Sb 5+
1.9
antimony
52 127.60
Te 2-
2.1
tellurium
53 126.90
I 1-
2.5
iodine
54 131.29
Xe 2.6
xenon

77 192.21
Ir 4+
2.2
iridium
78 195.08
Pt 2+
2.2
platinum
79 196.97
Au 1+
1.9
gold
80 200.59
Hg 2+
1.8
mercury
81 204.38
Tl 3+
1.8
thallium
82 207.20
Pb 4+
2.0
lead
83 208.98
Bi 5+
1.9
bismuth
84 209 [210]
Po 4+
2.0
polonium
85 210 [211]
At 1-
2.2
astatine
86 222
Rn 2.4
radon

104 [261]
Rf
105 [262]
Db
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Sg
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Bh
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Activity Series of Common Metals

Li reacts with water

Cs reacts with acid

Rb reacts with acid

K reacts with acid

Ba reacts with acid

Sr reacts with acid

Zn reacts with acid

Cr reacts with acid

Fe reacts with acid

Co reacts with acid

Ni reacts with acid

Sn reacts with acid